Application No. 10/533,580 Docket No.: 20241/0202878-US0 Amendment dated February 26, 2010

After Final Office Action of January 22, 2010

## LISTING OF THE CLAIMS

1. (Previously Presented) A dispersoid having metal-oxygen bonds which is obtained by mixing a metal compound having at least three hydrolyzable groups with a given amount of water in the absence of all members selected from the group consisting of an acid, a base and a dispersion stabilizer and at a given temperature, which dispersoid is characterized in that the given amount of water is at least 1.0 mole but less than 2.0 moles per mole of the metal compound, and the given temperature is a temperature below 0 °C, wherein a transmittance, expressed as a spectral transmittance measured at a dispersoid concentration of 0.5 wt % of oxide basis, at a quartz cell light path length of 1 cm, using the organic solvent as a control, and at a light wavelength of 550 nm. is 80% to 100%.

2. (Previously Presented) A dispersoid having metal-oxygen bonds which is obtained by mixing a metal compound having at least three hydrolyzable groups with a given amount of water in the absence of all members selected from the group consisting of an acid, a base and a dispersion stabilizer and at a given temperature, which dispersoid is characterized in that the given amount of water is at least 0.5 mole but less than 1.0 mole per mole of the metal compound, and the given temperature is a temperature below 0 °C, wherein a transmittance, expressed as a spectral transmittance measured at a dispersoid concentration of 0.5 wt % of oxide basis, at a quartz cell light path length of 1 cm, using the organic solvent as a control, and at a light wavelength of 550 nm, is 80% to 100%.

## 3-12. (Canceled)

13. (Previously Presented) A dispersoid having metal-oxygen bonds obtained by mixing, in the absence of all members selected from the group consisting of an acid, a base and a dispersion stabilizer and at a given temperature, a partial hydrolysate that is prepared by hydrolyzing a metal compound having at least three hydrolyzable groups in the absence of all members selected from the group consisting of an acid, a base and a dispersion stabilizer and that can be stably dispersed without aggregation in an organic solvent with an amount of water equal to at least 0.5 mole but less

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than 2 moles per mole of the metal compound minus the amount of water used to prepare the partial hydrolysate, which dispersoid is characterized in that the given temperature is a temperature below 0 °C.

14. (Original) The dispersoid having metal-oxygen bonds of claim 13 which is characterized in that the given temperature is a temperature of -20 °C or below.

15. (Original) The dispersoid having metal-oxygen bonds of claim 13 which is characterized in that the given temperature is at or below the temperature at which the metal compound begins to hydrolyze.

16-22. (Canceled)

23. (Original) The dispersoid having metal-oxygen bonds of claim 13 which is characterized by being obtained by, following mixture of the partial hydrolysate and the water at the given temperature, raising the temperature to the given temperature or above.

24. (Previously Presented) A dispersoid having metal-oxygen bonds which is obtained by mixing a metal compound having at least three hydrolyzable groups with a given amount of water in the absence of all members selected from the group consisting of an acid, a base and a dispersion stabilizer and at a given temperature, which dispersoid is characterized in that the given amount of water is a solution diluted with a hydrocarbon solvent other than an alcohol solvent, and an alcohol solvent, the diluted solution is added to the metal compound, and the given temperature is room temperature.

25. (Original) The dispersoid having metal-oxygen bonds of claim 24 which is characterized in that the given amount of water is at least 0.5 mole but less than 2.0 moles per mole of the metal compound.

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26. (Original) The dispersoid having metal-oxygen bonds of claim 24 which is characterized in that the water in the diluted solution has a concentration that is from 40% to 1% of the saturation solubility of water in a mixed solvent of the hydrocarbon solvent and the alcohol solvent.

- 27. (Previously Presented) A dispersoid having metal-oxygen bonds which is obtained by the addition, in the absence of all members selected from the group consisting of an acid, a base and a dispersion stabilizer, to a metal compound having at least three hydrolyzable groups, of at least 0.5 mole but less than 2 moles of water per mole of the metal compound, which dispersoid is characterized by having steps in which the water is added in divided portions at a given temperature, which steps include at least one step in which the given temperature is a temperature below 0 °C.
- 28. (Previously Presented) A dispersoid having metal-oxygen bonds which is obtained by the addition, in the absence of all members selected from the group consisting of an acid, a base and a dispersion stabilizer, to a metal compound having at least three hydrolyzable groups, of at least 0.5 mole but less than 2 moles of water per mole of the metal compound, which dispersoid is characterized by having steps in which the water is added in divided portions, at least 0.5 mole but less than 1 mole of the water per mole of the metal compound being added in a first water addition step.
- 29. (Original) The dispersoid having metal-oxygen bonds of claim 28 which is characterized by having, after the first water addition step, a step in which the rest of the required amount of water is added at a given temperature, the given temperature being a temperature below 0 °C.

30-42. (Canceled)

43. (Previously Presented) A dispersoid having metal-oxygen bonds which is characterized by dispersing stably without aggregation in an organic solvent in the absence of all members selected from the group consisting of an acid, a base and a dispersion stabilizer, and by having an average particle size in a range of 1 to 20 nm, wherein a transmittance, expressed as a spectral transmittance measured at a dispersoid concentration of 0.5 wt % of oxide basis, at a quartz cell light path length

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of 1 cm, using the organic solvent as  $\underline{a}$  control, and at a light wavelength of 550 nm, is 80% to

100%.

44. (Previously Presented) The dispersoid having metal-oxygen bonds of claim 43 which is

characterized by being monodispersed with a particle size distribution in a range of 0 to 50 nm.

45-53. (Canceled)

54. (Withdrawn) A metal oxide film which is characterized by being formed by coating or

spraying, and by having a smooth film surface.

55. (Withdrawn) The metal oxide film of claim 54 which is characterized by being formed by

drying at 200 °C or below.

56. (Withdrawn) The metal oxide film of claim 54 which is characterized in that the film surface

has an average roughness of 10 nm or less.

57. (Canceled)

58. (Withdrawn) A metal oxide film which is characterized by being formed on a plastic

substrate and by having a carbon content, expressed as an atomic ratio, of 10% or less.

59-64. (Canceled)

65. (Withdrawn) A monomolecular film which is characterized by being obtained by forming a

metal oxide film having a smooth surface on a substrate, then contacting the metal oxide film with a

metallic surfactant having at least one hydrolyzable group.

66. (Canceled)

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67. (Withdrawn) A monomolecular film characterized by being obtained by using a dispersoid having metal-oxygen bonds that is stably dispersed without aggregation in an organic solvent in the absence of at least one selected from the group consisting of an acid, a base and a dispersion stabilizer, or using a solution containing the dispersoid, to form a metal oxide film on a substrate, then contacting the metal oxide film with a metallic surfactant having at least one hydrolyzable group.

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68-70. (Canceled)